

Supplementary Materials

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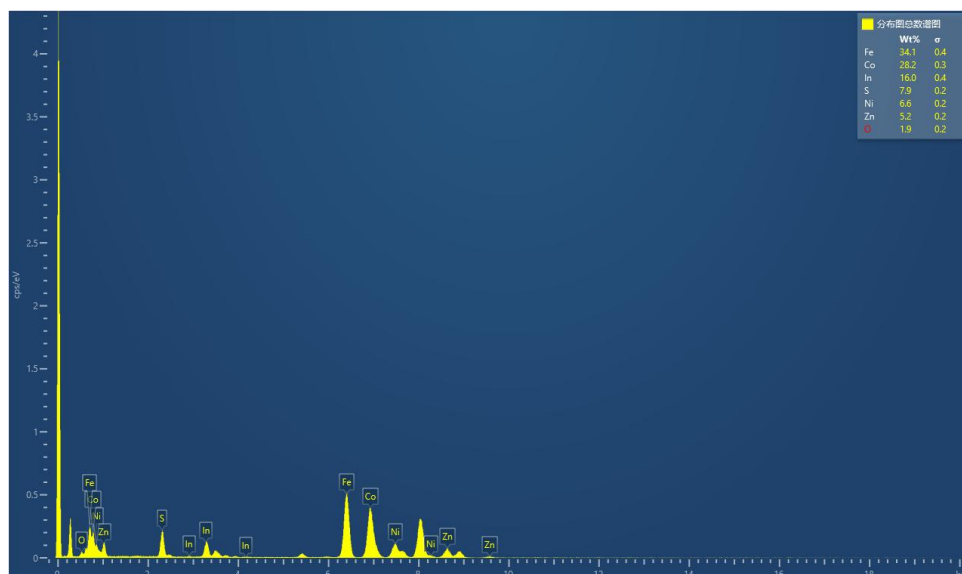
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Characterization and Measurement Details

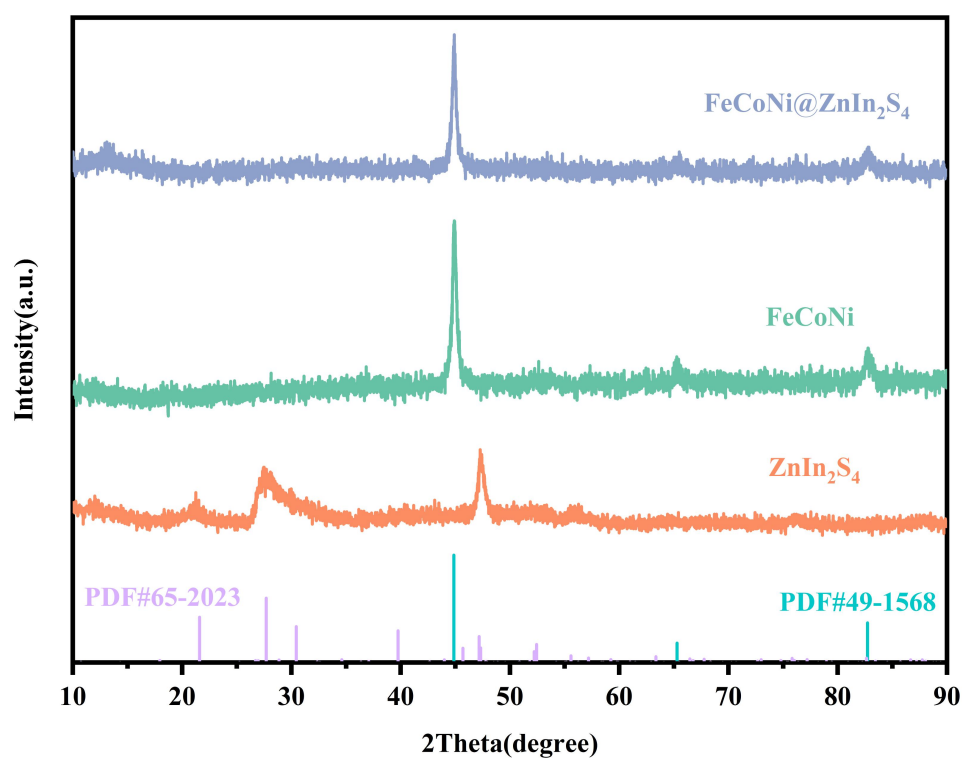
The crystal structure of the samples was analyzed by X-ray diffraction (XRD, Brucker, D8 ADVANCED) with Cu-K α radiation ($\lambda=1.5406 \text{ \AA}$, 40 kV, 40 mA). The morphology of the samples was observed using field emission scanning electron microscope (FESEM, HITACHI, S4800) with a working voltage and current of 5.0 kV and 10 μA , respectively. The JEOL JEM-2100F transmission electron microscope (TEM) was used to analyse the material morphology, the operating voltage, current and current density of 200 kV, 212 μA and 511.8 pA/cm², respectively. The XPS spectra in this study were obtained using a Thermo Scientific ESCALAB 250Xi spectrometer equipped with a monochromatic X-ray source (Al K α). During X-ray photoelectron spectra acquisition, the energy resolution $\leq 0.5 \text{ eV}$, pass energy = 10 eV. Calibrate the main peak position of the C 1s peak for contaminated carbon to 284.8 eV, then use the Shirley background subtraction method, with the subtraction operation performed in Avantage. Based on the expected chemical states of the elements and a standard binding energy database, preliminary peak positions were set. The peak positions, peak areas, and half-widths were optimised using the non-linear least squares fitting function in Avantage. Finally, the fitting quality was assessed by visually inspecting the fit of the curve to the raw data and the randomness of the residual plot.

The mixtures of ZnIn₂S₄, FeCoNi, ZnIn₂S₄-500, ZnIn₂S₄-600, ZnIn₂S₄-700, FCNZ, FCNZ-500, FCNZ-600, and FCNZ-700 composites with 60 wt% paraffin was compacted into coaxial ring of 7 mm outer diameter and 3.04 mm inner diameter to assess EM wave absorption performance.

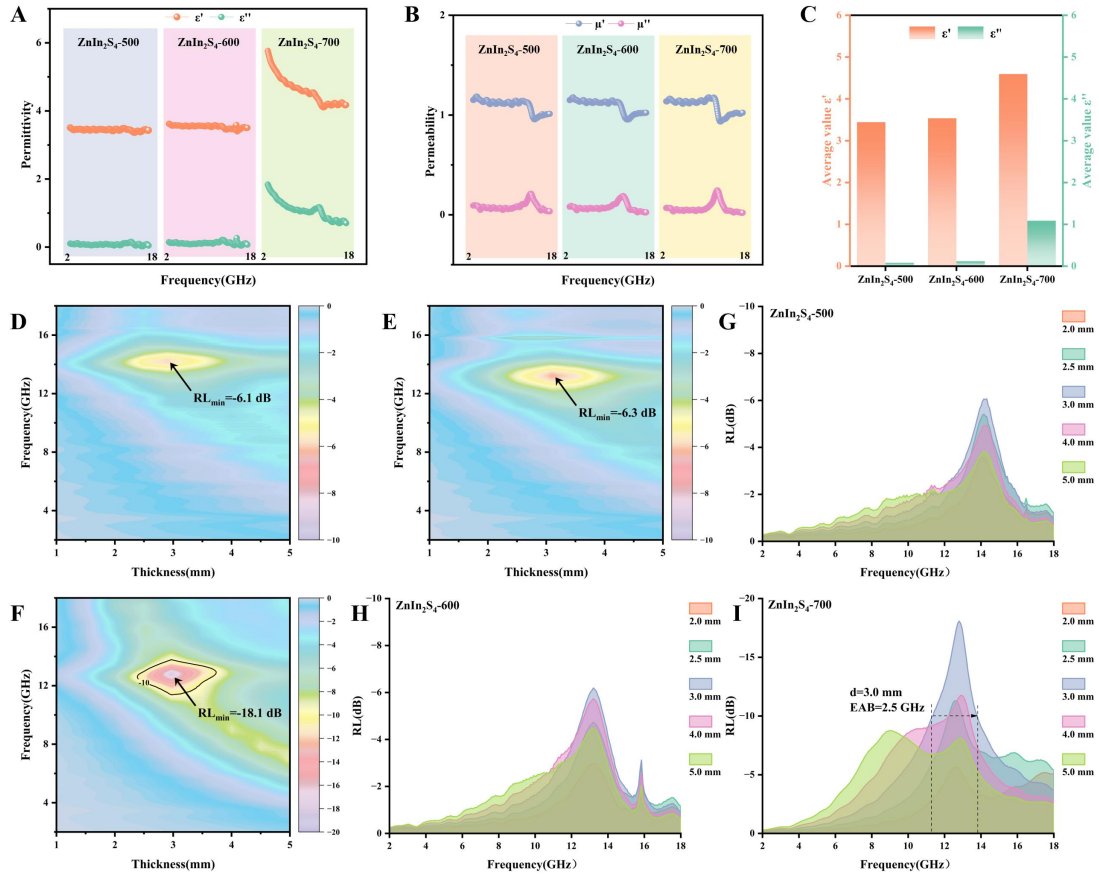
Supplementary Figures and Tables



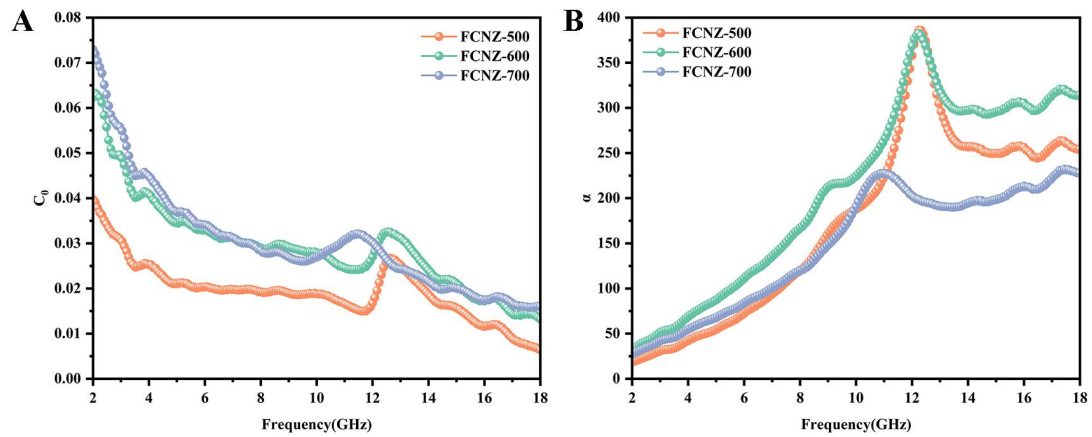
Supplementary Figure 1. EDS spectra of FCNZ-600.



Supplementary Figure 2. The XRD pattern of synthesized ZnIn₂S₄, FeCoNi, and FeCoNi@ZnIn₂S₄.



Supplementary Figure 3. (A) The complex permittivity, (B) complex permeability, (C) average complex permittivity, (D-F) 3D and (G-I) 2D reflection loss diagrams of $\text{ZnIn}_2\text{S}_4\text{-}500$, $\text{ZnIn}_2\text{S}_4\text{-}600$, and $\text{ZnIn}_2\text{S}_4\text{-}700$.



Supplementary Figure 4. (A) C_0 -f curve of composites, (B) α -f curve of composites.

Supplementary Table 1. Microwave absorption performance of core-shell structured absorbers in previous references and this work.

Sample	EAB (GHz)	RL _{min} (dB)	Thickness (mm)	Reference
Fe ₃ O ₄ @C	1.6	-47	2	[1]
Fe ₃ O ₄ @ppy	2.4	-46	5	[2]
Ni@C@ZnO	3.3	-55.8	2.50	[3]
	4.10	-27	2.00	
Ni@ppy	3.8	-48	5	[4]
H-Fe ₃ O ₄ @C	5.36	-51.85	2.1	[5]
Fe/Fe ₃ C@SiO ₂ @C	3.74	-48.68	1.24	[6]
Fe ₃ O ₄ @SiO ₂ @MoSe ₂	4.96	-51.8	1.8	[7]
Fe ₃ Al@PPy	3.44	-40.53	2.0	[8]
NiFe ₂ O ₄ @PPy	5.2	-56.25	2.24	[9]
	7.12	-	2.6	
C@NiCo ₂ O ₄ @Fe ₃ O ₄	2.1	-43	3.4	[10]
FCNZ-600	4.91	-52.4	1.9	This Work
	6.08	-34.7	1.53	This Work

REFERENCES

1. Adebayo, L.L. A Simple Route to Prepare Fe₃O₄@C microspheres as electromagnetic wave absorbing material. *J. Mater. Res. Technol.* **2021**, *12*, 1552-1563. DOI: 10.1016/j.jmrt.2021.03.094
2. Tian, X.; Xu, X.; Bo, G. A 3D flower-like Fe₃O₄@PPy composite with core-shell heterostructure as a lightweight and efficient microwave absorbent. *J. Alloys Compd.* **2022**, *923*, 166416. DOI:10.1016/j.jallcom.2022.166416
3. Wang, L.; Yu, X.; Li, X. MOF-derived yolk-shell Ni@C@ZnO schottky contact structure for enhanced microwave absorption. *Chem. Eng. J.* **2020**, *383*, 123099. DOI:10.1016/j.cej.2019.123099
4. Zhang, N.; Li, J.; Men, X. Cleaning synthesis of core-shell structured Ni@PPy composite as excellent lightweight electromagnetic wave absorber. *J Mater Sci: Mater Electron.* **2020**, *31*, 1483–1490. DOI:10.1007/s10854-019-02663-5
5. Guo, J.; Sun, Y.; Li, X. Carbonization tuned core-shell Fe₃O₄@C nanostructures with enhanced electromagnetic wave absorption. *Adv Mater. Interfaces.* **2025**, 2500075. DOI:10.1002/admi.202500075
6. Deng, Z.; Wang, L.; Peng, B. Optimization of electromagnetic wave absorption properties by formation of magnetoelectric synergistic effect of COF-derived carbon composite Fe/Fe₃C. *Chem. Eng. J.* **2025**, *505*, 159457. DOI:10.1016/j.cej.2025.159457
7. Li, Q.; Yu, G.; Ye, M. Rational design of flower-like core-shell Fe₃O₄@SiO₂@MoSe₂ composites for high performance electromagnetic wave absorption. *J. Mater. Sci: Mater Electron.* **2023**, *34*, 1723. DOI:10.1007/s10854-023-11098-y
8. Shen, Y.; Song, P.; Zhang, F. Core-shell Fe₃Al@polypyrrole composites with tunable shell thickness and enhanced microwave absorption performance. *Colloid. Surface. A.* **2024**, *682*, 132814. DOI:10.1016/j.colsurfa.2023.132814
9. Shu, R.; Yun, K.; Liu, X. Fabrication of core-shell nickel ferrite@polypyrrole composite for broadband and efficient electromagnetic wave absorption. *Compos. Part A-Appl. S.* **2025**, *188*, 108558. DOI:10.1016/j.compositesa.2024.108558
10. Wei, S.; Wang, X.; Zhang, B. Preparation of hierarchical core-shell C@NiCo₂O₄@Fe₃O₄ composites for enhanced microwave absorption performance. *Chem. Eng. J.* **2017**, *314*, 477–487. DOI:10.1016/j.cej.2016.12.005