

Supplementary Materials

Topological insulator modified nickel oxide nanoflowers for low-temperature hydrogen sulfide gas sensing

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First-principles calculation methods

In this work, we utilized first-principles calculations based on density functional theory (DFT) to elucidate the gas sensitivity of NiO_x and $\text{Bi}_2\text{Te}_3@\text{NiO}_x$. The CASTEP software package was employed to determine the adsorption energy and charge density differences of these materials. The Perdew-Burke-Ernzerhof (PBE) functional, coupled with D2 dispersion correction, was selected to account for exchange-correlation effects. Ultrasoft pseudopotentials were utilized to model electron-ion interactions, while a plane-wave basis set was applied for the expansion of the Kohn-Sham orbitals. The Monkhorst-Pack K-point grid was sampled at densities of $2\pi \times 0.04 \text{ \AA}^{-1}$ and $2\pi \times 0.02 \text{ \AA}^{-1}$ for geometry optimizations and electronic structure calculations, respectively. A kinetic energy cut-off of 400 eV was established for these calculations. The structural optimization was considered complete when the total energy convergence reached 10^{-5} eV , and the force on each atom was below $0.01 \text{ eV \AA}^{-1}$. To avoid interactions between adjacent periodic images, a $20 \text{ \AA}$ vacuum space was incorporated into the model. Furthermore, the band structure was computed using the VASP software, with the inclusion of spin-orbit coupling (SOC) to provide a comprehensive analysis.

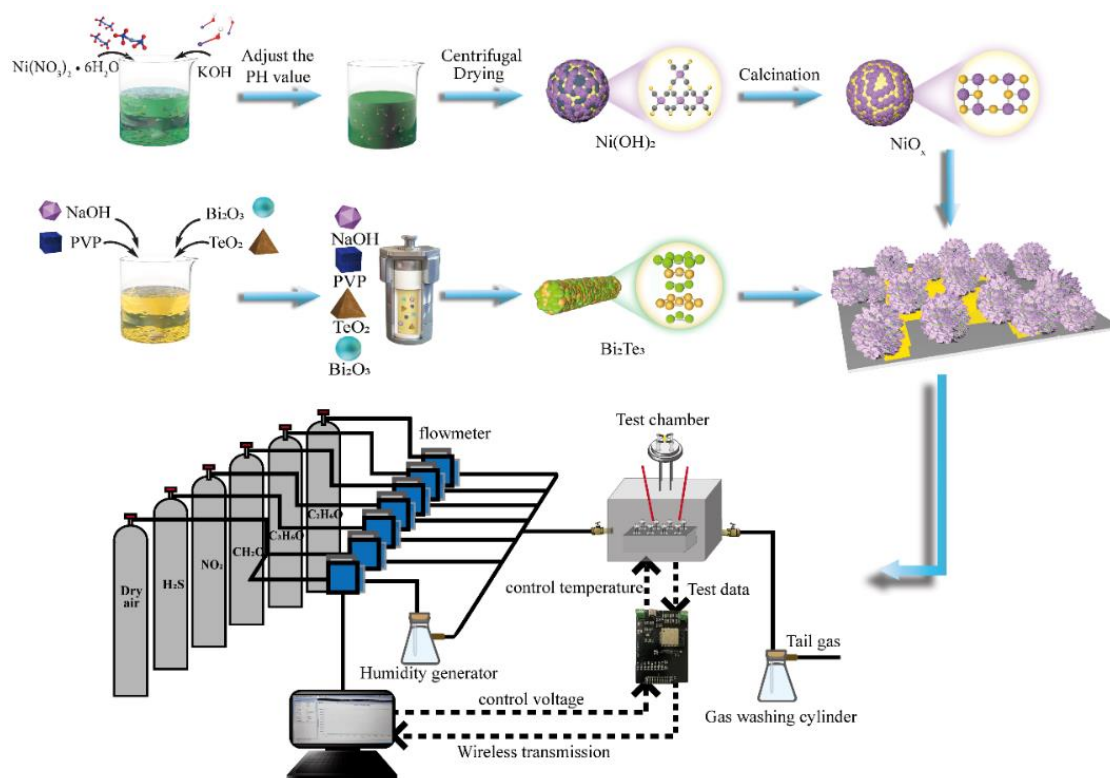


Figure S1. Schematic diagram of the synthesis of $\text{Bi}_2\text{Te}_3@\text{NiO}_x$ and the testing system.

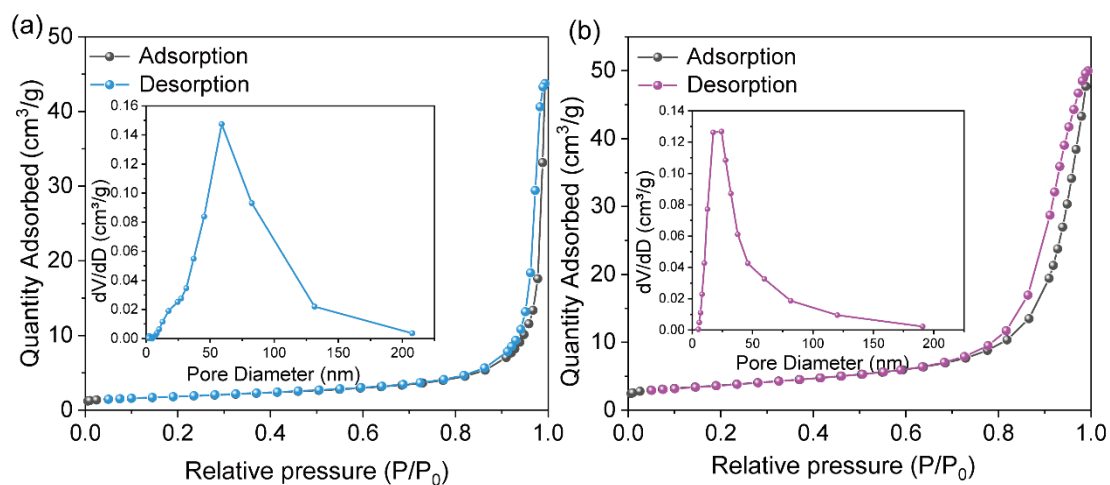


Figure S2. N_2 adsorption-desorption isotherms and pore size distribution curves of pure NiO_x (a) and 10 wt% $\text{Bi}_2\text{Te}_3@\text{NiO}_x$ (b).

Table S1. The comparison among our work and recent H₂S gas sensors

Materials	T (°C)	Resp.	Con. (ppm)	LOD (ppm)	Ref.
SnO ₂	160	5	9	3	[1]
NiO	175	5.8	100	3	[2]
Nb ₂ O ₅ @ZnO	400	6.1	20	1	[3]
WO ₃	300	1.84	100	-	[4]
CuO-WO ₃	300	6.72	100	-	[4]
rGO/CuO	300	11.7	10	1	[5]
SnO ₂ /RGO: Pt	200	3.7	40	1	[6]
Bi ₂ Te ₃ @NiO	90	11.81	10	0.2	This work

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