

Supplementary Materials

Reconstruction-mediated interfacial water engineering in Cu-Mo electrocatalyst for highly efficient and durable water splitting

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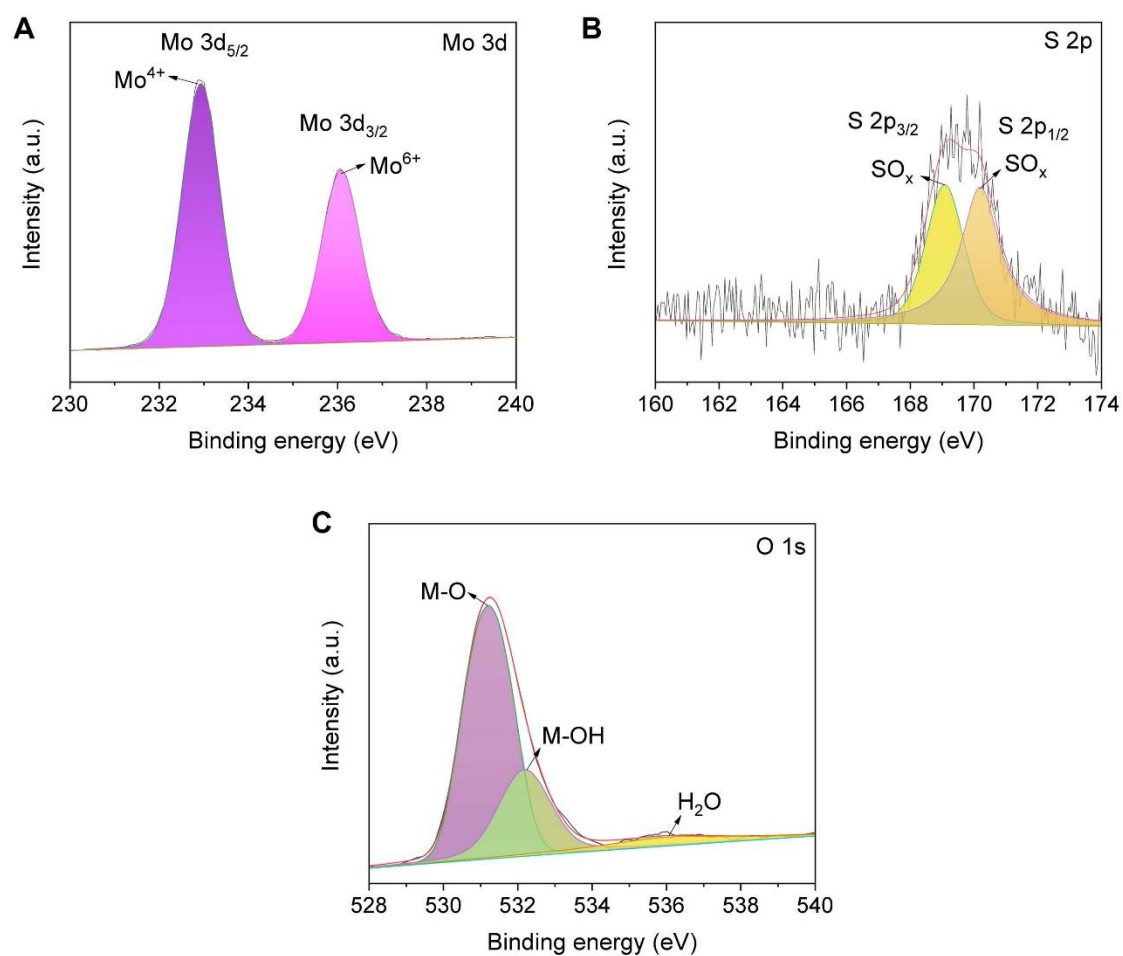
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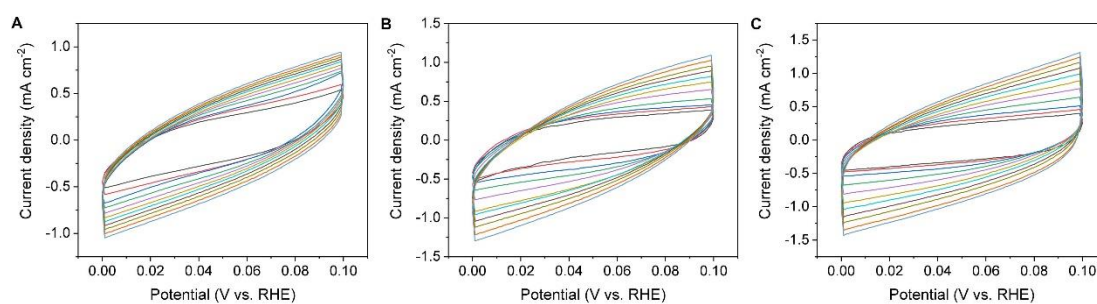
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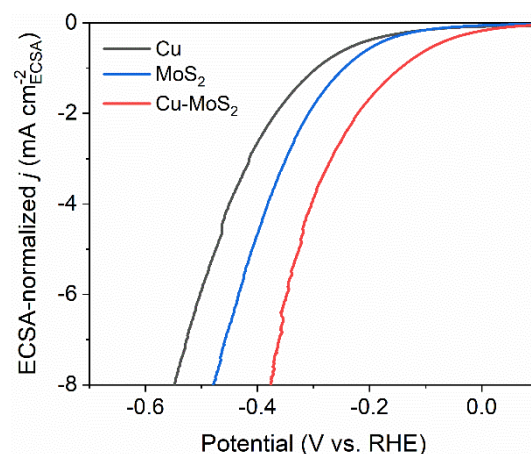
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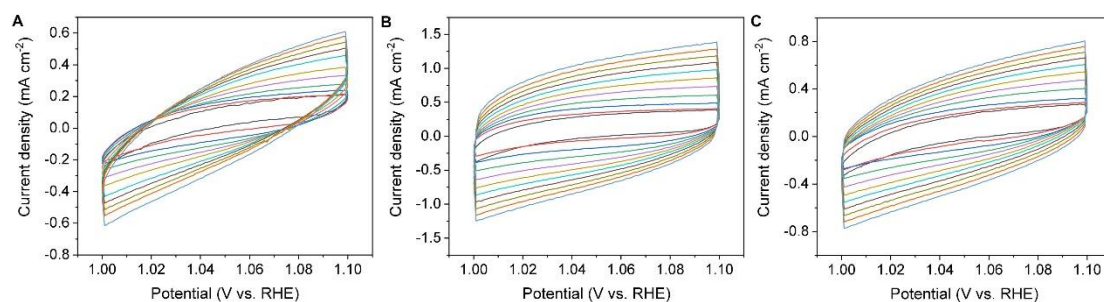
Supplementary Figure 1. XPS spectra of (A) Mo 3d, (B) S 2p, and (C) O 1s for MoS₂/MoO₃ nanosheets.



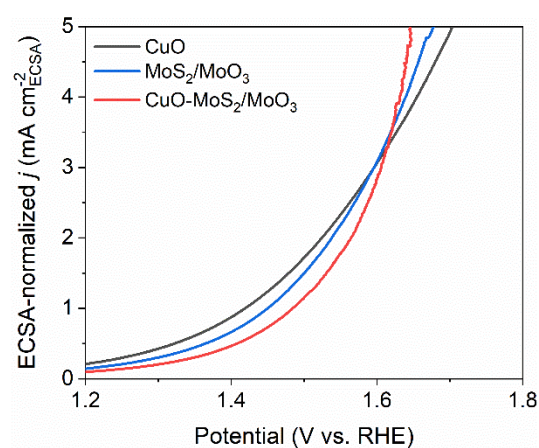
Supplementary Figure 2. CV curves of (A) Cu, (B) MoS₂, and (C) Cu-MoS₂ for HER under different scan rates from 10 to 200 mV s⁻¹.



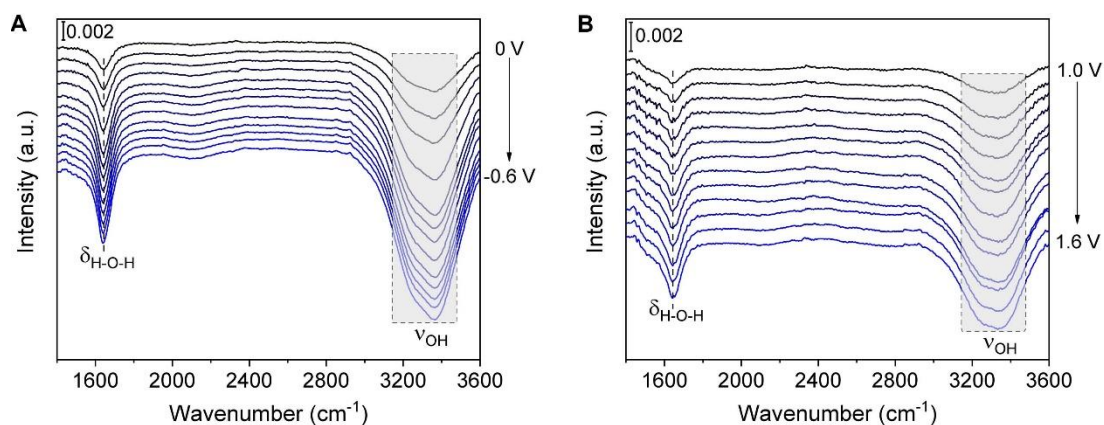
Supplementary Figure 3. ECSA-normalized current density of Cu-MoS₂ and control catalysts for HER.



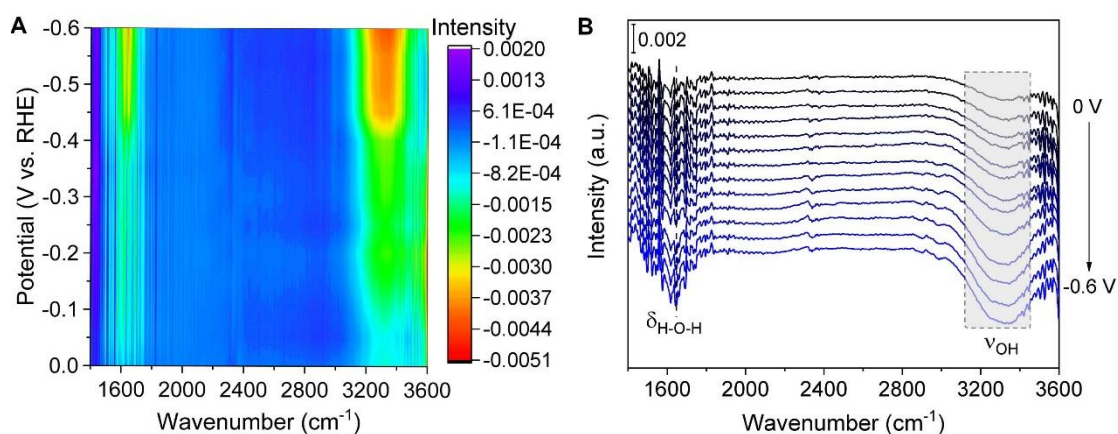
Supplementary Figure 4. CV curves of (A) CuO, (B) MoS₂/MoO₃, and (C) CuO-MoS₂/MoO₃ for OER under different scan rates from 10 to 200 mV s⁻¹.



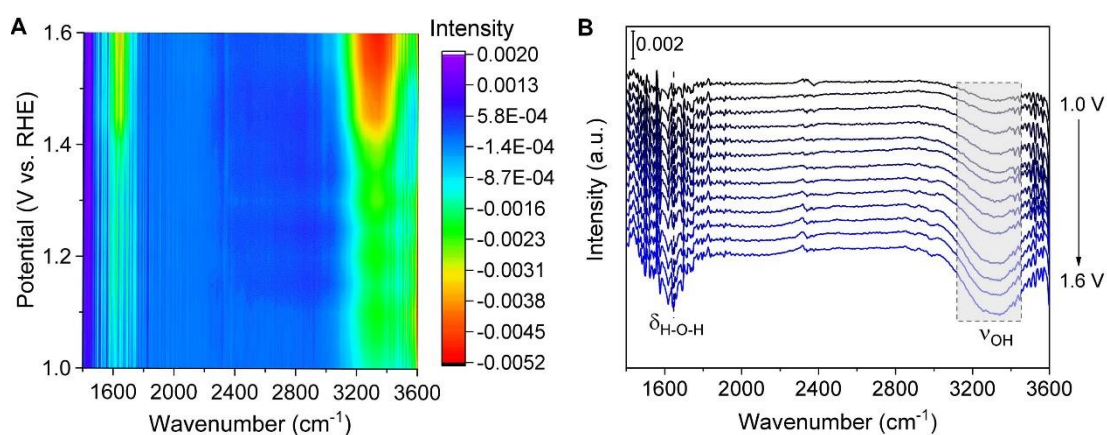
Supplementary Figure 5. ECSA-normalized current density of CuO-MoS₂/MoO₃ and control catalysts for OER.



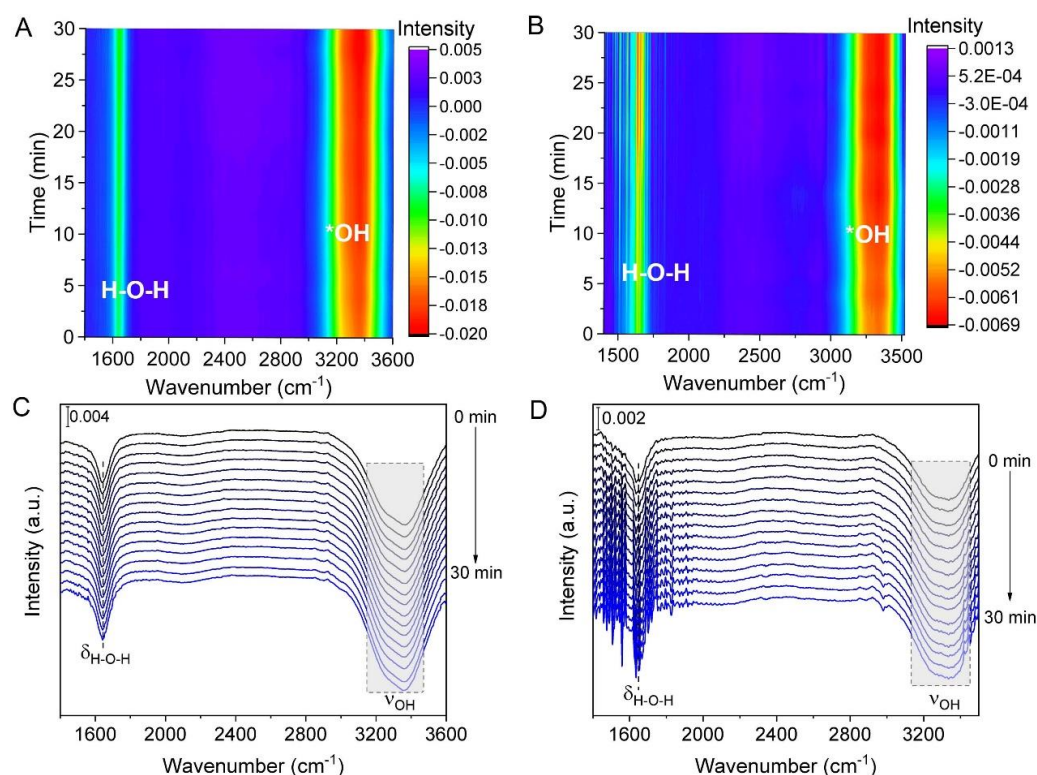
Supplementary Figure 6. In situ FTIR spectra of (A) Cu-MoS₂ for HER under the potential from 0 to -0.6 V and (B) CuO-MoS₂/MoO₃ for OER under the potential from 1.0 to 1.6 V.



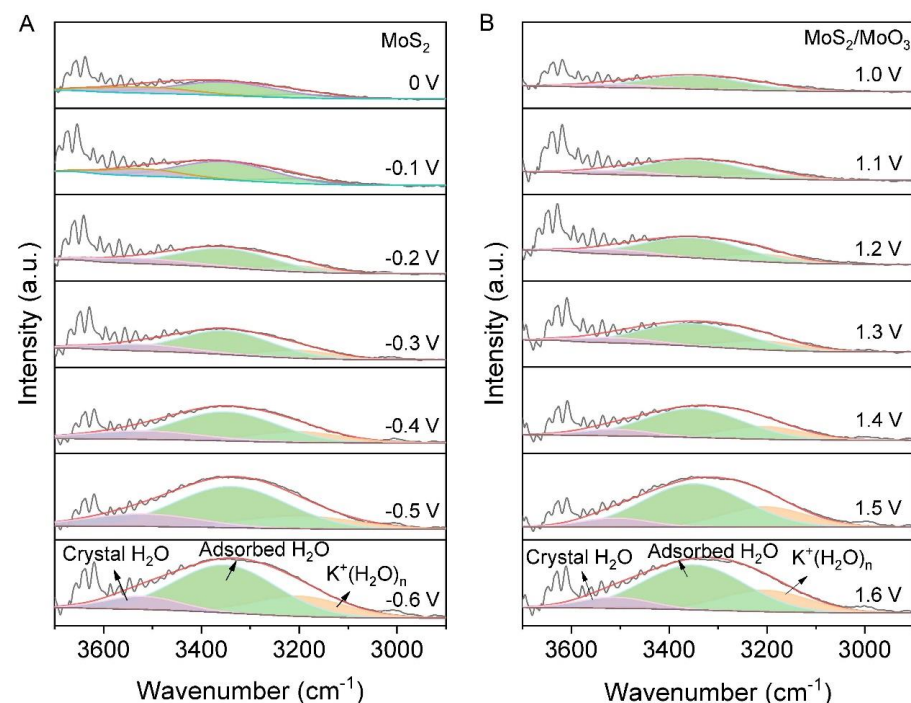
Supplementary Figure 7. In situ FTIR spectra of MoS₂ for HER under the potential from 0 to -0.6 V.



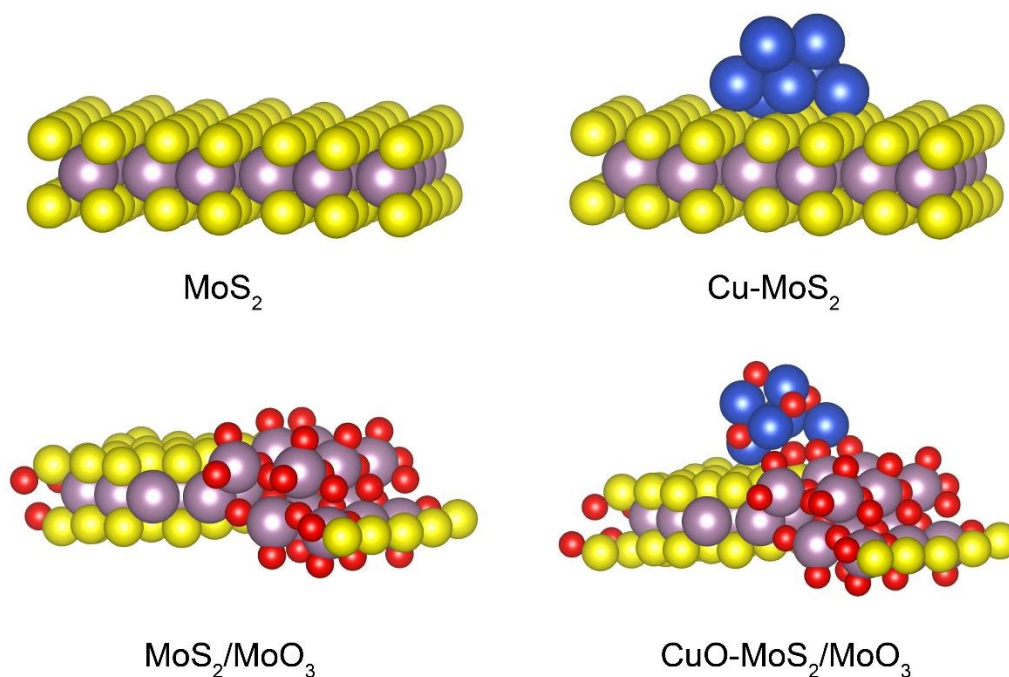
Supplementary Figure 8. In situ FTIR spectra of MoS₂/MoO₃ for OER under the potential from 1.0 to 1.6 V.



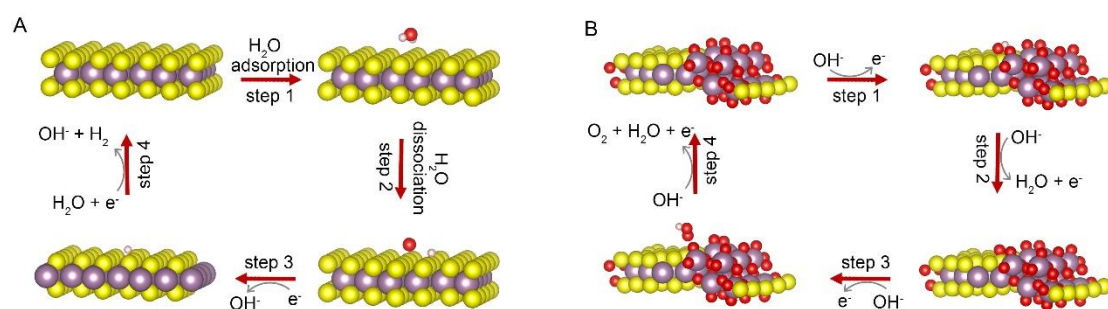
Supplementary Figure 9. Time-dependent in situ FTIR spectra of (A, C) Cu-MoS₂ for HER under a potential of -0.5 V for 30 min, and (B, D) CuO-MoS₂/MoO₃ for OER under a potential of 1.5 V for 30 min.



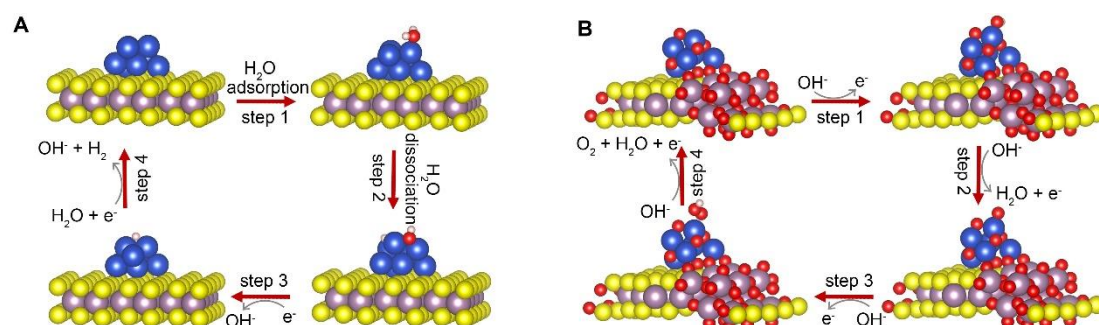
Supplementary Figure 10. The Gaussian fits of three O-H stretching modes on (A) MoS₂ and (B) MoS₂/MoO₃.



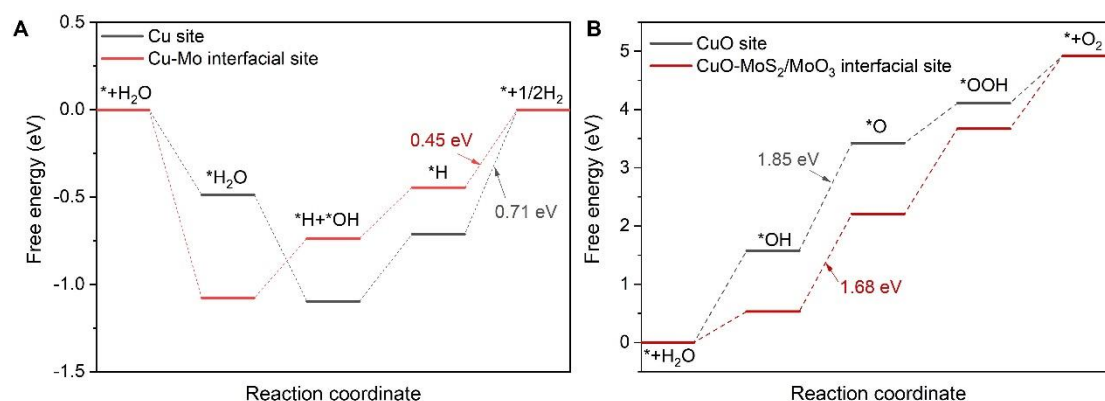
Supplementary Figure 11. Structural models of MoS_2 , Cu-MoS_2 , $\text{MoS}_2/\text{MoO}_3$, and $\text{CuO-MoS}_2/\text{MoO}_3$.



Supplementary Figure 12. Reaction pathways of (A) HER process on MoS_2 and (B) OER process on $\text{MoS}_2/\text{MoO}_3$.



Supplementary Figure 13. Reaction pathways of (A) HER process on Cu-MoS_2 at Cu site and (B) OER process on $\text{CuO-MoS}_2/\text{MoO}_3$ at CuO site.



Supplementary Figure 14. Gibbs free energy diagram for (A) HER process on Cu-MoS₂ at Cu site and Cu-Mo interfacial site, and (B) OER process on CuO-MoS₂/MoO₃ at CuO site and CuO-MoS₂/MoO₃ interfacial site.

Supplementary Table 1. ECSA-normalized activity, turnover frequency (TOF), and mass activity for HER.

Catalysts	ECSA	$j_{\text{ECSA}}^{\text{HER}}$ (mA cm ⁻² _{ECSA})		TOF (s ⁻¹)		Mass activity (A g ⁻¹)	
		$\eta=100$ mV	$\eta=500$ mV	$\eta=100$ mV	$\eta=500$ mV	$\eta=100$ mV	$\eta=500$ mV
Cu	77.75	0.153	5.92	0.48	18.5	23.8	920.6
MoS ₂	109.25	0.155	8.97	0.48	28.0	34.0	1969.8
Cu-MoS ₂	143.25	0.611	8.94	1.91	27.9	175.2	2562.3

Supplementary Table 2. ECSA-normalized activity, turnover frequency (TOF), and mass activity for OER.

Catalysts	ECSA	$j_{\text{ECSA}}^{\text{OER}}$ (mA cm ⁻² _{ECSA})		TOF (s ⁻¹)		Mass activity (A g ⁻¹)	
		$\eta=200$ mV	$\eta=500$ mV	$\eta=200$ mV	$\eta=500$ mV	$\eta=200$ mV	$\eta=500$ mV
CuO	55.75	1.078	5.594	0.168	0.870	120.2	623.7
MoS ₂ /MoO ₃	104.50	0.847	6.731	0.132	1.052	176.9	1406.8
CuO-MoS ₂ /MoO ₃	204.25	0.612	5.981	0.095	0.932	250.0	2442.8

Supplementary Table 3. Comparison of the performance of the Cu-MoS₂||CuO-MoS₂/MoO₃ electrolyzer with reported electrolyzers.

Electrolyzer		Temperature (°C)	Cell voltage (V) @ 1.0 A cm ⁻²	Stability (h)	Ref.
Cathode	Anode				
Cu-MoS ₂	CuO-MoS ₂ /MoO ₃	80	1.77	500	This work
Ru _{8%} -Co ₃ O ₄	Ru _{8%} -Co ₃ O ₄	85	2.06	120	[1]
MCHEP	MCHEP	80	1.83	400	[2]
Ru,Mn-NiCoP	Ru,Mn-NiCo LDH	85	2.0	80	[3]
Ir _{SA} -NiFe LDH/NiMo	Ir _{SA} -NiFe LDH/NiMo	60	1.72	120	[4]
FeO _x @Co _{0.75} Fe _{0.25} P	FeO _x @Co _{0.75} Fe _{0.25} P	60	1.73	120	[5]
NiMn-OPSe	NiMn-OPSe	70	2.07	500	[6]
NiFe@Ni/Fe-MnOOH	NiFe@Ni/Fe-MnOOH	25	1.953	200	[7]
Ni(OH) ₂ @Pt ^{δ-} -NC	NiFe-LDH	25	1.78	200	[8]
Pt-C core-shell @h-MoS ₂ /GNF	NiCo ₂ O ₄	60	2.18	15	[9]
Y-NiMo/MoO _{2-x}	NiFe-LDHs	80	1.78	24	[10]
NiFeCu	NiFeCu	25	2.13	12	[11]
Ir-Ni(OH) ₂ /NF	Ir-Ni(OH) ₂ /NF	60	1.64	225	[12]
HS Co ₃ S ₄	Mo-CoOOH	60	1.78	1000	[13]
Ni ₂ P@FeP@Co ₂ P	Ni ₂ P@FeP@Co ₂ P	85	1.95	500	[14]
Pt/C	NiFe-LDH-PTA	60	1.93	170	[15]
Co ₂ Mo ₃ O ₈ /MoO ₂ /NF	FeMoOOH/NF	65	1.76	75	[16]
P-Os/NiFe	RuO ₂ /NiFe	60	2.02	100	[17]
Er _{0.1} -CoP/NiCoP@NF	Ce-Fe _x P/Ni ₃ P@NF	50	1.783	1000	[18]
Ru-Ru ₂ P/V ₂ CT _x	RuO ₂	75	1.80	550	[19]

Cu-MoNi ₄ @NF	NiFeOOH	60	1.74	1000	[20]
N-NiMo/MoO ₂	NiFe	80	1.87	120	[21]
Cu/Ni@MoO ₂	NiFeOOH	60	1.69	1000	[22]
Pt/C	Mo-doped Ni ₃ S ₂ /FeNi ₂ S ₄	60	1.66	200	[23]
NiSe	r-NiMoSe _x	80	1.70	190	[24]
Ni@Ti-MoO _x	NiFe LDH	60	1.72	100	[25]
Cr-NiMo/MoO ₂	Ni ₃ S ₂ -NiFe LDH	80	1.74	30	[26]
Mo ₂ C/NiMo	Mo ₂ C/FeNiMo	60	~1.78	100	[27]
Ru-MoN ₄ -NC	NiFe LDH	60	1.72	1200	[28]
NiRu-MOF	FeNi-MOF	25	2.00	350	[29]
Os SA/NiCu	NiFe LDH	80	1.76	500	[30]

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